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Chemistry Principles of Modern Chemistry In some reactions there is a competition between kinetic control and thermodynamic control over product yields. Suppose compound

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A can undergo two elementary reactions to stable products:  $A \xrightleftharpoons[k_{-1}]{k_1} B$  or  $A \xrightleftharpoons[k_{-2}]{k_2} C$ . For simplicity we assume first-order kinetics for both forward and reverse reactions.

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