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Note taking guide: Episode 302. Nucleons. Atomic number. Neutral atom. Isotope. particles in the nucleus of atom. number of protons in the nucleus of an atom. number of protons (+) = number of electrons (-). atoms of an element that have different numbers of neutrons.

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Season 1 Episode 302 | 24m 14s Atomic Theory: Structure of the Atom: Students use the terms isotope, atomic number, mass number, and atomic mass to determine the number of subatomic particles in ...

Chemistry 6 Physics | Chemistry 302: The Structure of the ...

Episode 303. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. LStavar5. Terms in this set (7) Bohr's Energy Levels. 1. electrons have certain energy levels 2. low energy levels- closer to nucleus 3. high energy levels- farther from nucleus 4. ground state- electron in lowest energy level possible.

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